



HI-60001-02 PH 1.000 MILLESIMAL BUFFER SOLUTION, 25 X 20 ML SACHETS

Reference: HI-60001-02

Designed and manufactured for those where accuracy really is vital, Millesimal buffer solutions are produced to a premium quality for customers who require a greater resolution of 0.001 for pH measurements.

- ✓ **Accuracy: +/- 0.002 @ 25°C**
- ✓ **Certificate of Analysis**
- ✓ **25 x 20 ml**

FULL DESCRIPTION

To achieve the more accurate results, Millesimal Buffer solutions are available in 14 values: pH1.000, pH1.679, pH2.000, pH3.000, pH4.010, pH6.000, pH6.862, pH7.010, pH8.000, pH10.010, pH11.000, pH12.000, pH12.450, pH13.000, available in 500ml, 1 litre, and handy 20ml sachets in packs of 25 for on-site or field use. The sachet single dose is safe, easy to carry and remains fresh.

Millesimal buffer solutions come with a certificate of analysis on purchase.

HI-60001-02 pH1.000 Millesimal Buffer Solution, 25 x 20ml sachets

HI-60001-02 is a premium quality pH 1.000 Millesimal buffer solution, National Institute of Standards and Technology (NIST) traceable and supplied with a Certificate of Analysis. The buffer is ideal for users expecting a highly-acidic reading.

The single dose sachets are really handy with each sealed sachet holding just the right amount of solution. Each time a meter or electrode is calibrated using a Hanna sachet, it is like using a newly opened bottle of solution!

Benefits of individual packaging

- Ideal for users who do not calibrate frequently
- Ideal for field use due to portability
- Every sachet is as fresh as the day it was packaged
- Light block packaging prevents oxidation from UV light that could alter the buffer value.

Buffer solutions should be stored at 25 C, away from direct sunlight for accurate readings.

To correctly maintain your meter/electrode, we highly recommend use of a Hanna cleaning solution, as well as a buffer, for maximum performance every time.

Hanna buffer solutions are prepared, ready-to-use, according to precise formulas and standardised in accordance with the National Institute of Standards and Technology (NIST) to ensure quality and accuracy. All solutions show a batch number, an expiry date, and the correlation table between pH and temperature. All bottled solutions are sealed with an air tight, tamper-proof seal to prevent any Carbon Dioxide from entering and causing a change in the pH buffer value.

Browse the Hanna complete range of solutions

- pH buffer solutions
- Standard solutions for conductivity, Total Dissolved Solids (TDS), turbidity, salinity and Ion-Selective Electrode (ISE) calibration
- Oxidation Reduction Potential (ORP) test and pre-treatment solutions
- Reference fill solutions for refillable electrodes
- General and specific cleaning solutions for electrodes
- Solutions for electrode maintenance and storage

- Solutions for sample preparation

Solutions are available in many sizes ranging from 20ml sachets for small users to 3.78L (1 gallon) containers for large quantity usage in laboratories.

For safety and traceability, all Hanna solutions are provided with a label showing the batch number and expiry date. A certificate of analysis is available on request in line with NIST standards.

Remember:

- Calibration only counts when using fresh solutions and properly cleaned electrodes every time.
- Routine maintenance will ensure accurate readings while extending the life of the electrode.

Solutions are available in a variety of sizes. Check whether the solution you require is available in 1 gallon, 500ml, or 230ml bottles, or individual 20ml sachets.



WolfLabs

Pricing on any accessories shown can be found by keying the part number into the search box on our website.

The specifications listed in this brochure are subject to change by the manufacturer and therefore cannot be guaranteed to be correct. If there are aspects of the specification that must be guaranteed, please provide these to our sales team so that details can be confirmed.

www.wolflabs.co.uk

Tel : 01759 301142

Fax : 01759 301143

sales@wolflabs.co.uk

Please contact us if this literature doesn't answer all your questions.